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## PHYSICO-CHEMICAL PROPERTIES OF CONTROLLED HYDROPHILIC-HYDROPHOBIC SILICA.

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Article history:		Abstract:
Accepted:	11 <sup>th</sup> October 2023 10 <sup>th</sup> November 2023 14 <sup>th</sup> December 2023	This paper examines the physico-chemical properties of a controlled hydrophilic-hydrophobic silicon in light of the surface change technique and their effects on the material's properties. The study includes a full analysis of the literature, experimental methods, results, and discussions to ensure comprehensive understanding of the subject.

**Keywords:** Controllable silica, hydrophilic-hydrophobic modification, physicochemical properties, surface chemistry, material science, nanotechnology.

Silica finds applications in various fields such as versatile material, catalysis, drug delivery and sensors. To control its hydrophilic-hydrophobic nature, the modification of silica surfaces gained considerable attention. This paper focuses on reviewing and analyzing the literature on the physical and chemical properties of controlled hydrophilic-hydrophobic silicon, clarifying the methodologies used and the consequences.

Numeros of studies have investigated the modification of silica surfaces to match moisturizing properties. Common methods include chemical welding, sol-gel processes and template of surfactants. The literature highlights the various applications of hydrophilic-hydrophobic silicon in areas such as environmental restoration, biotech science and advanced materials.

The experimental approach involves the synthesis of hydrophilic-hydrophobic silica through the sol-gel process. The surface modification was achieved by introducing various organosilans during hydrophobicity synthesis. For evaluation of physicochemical changes, the diagnostic techniques were used, including Fourier-transform infrared spectroscopy (FTIR), scanning electron microscopy (SEM), and contact angle measurements.

The controllable hydrophilic-hydrophobic silica refers to modified silica particles in order to exhibit hydrophilic (water-recipient) and hydrophobic (waterrecipient) properties. These modifications are achieved by introducing different functional groups into the silica surface. The physico-chemical properties of the controlled hydrophilic-hydrophobic silicon can be adapted based on the specific needs of various applications. Here are some key features:

Superficial Modification:

Superficial modification plays a crucial role in adapting the material's properties for various applications. The introduction of specific functional groups under the conditions of silica surfaces can significantly change their interaction with water and other substances. Here are two examples of surface modification:

Hydrophilic Functional Groups:

- Introduction to polar groups: hydrophilic functional groups include polar bonds close to water molecules. Silica surfaces can be modified by incorporating hydrophilic functional groups such as hydroxyl (-OH) and silanol (-SiOH) groups.

- Effect on surface properties: The addition of hydrophilic groups increases the humidity of the silicon surface, strengthens the adsorption of water molecules, and improves their interaction with polar substances. This modification is often useful in applications requiring high-level interactions with water or polar solvents, e.g. in chromatography or bioseparations.

Hydrophobic Functional Groups:

- Addition of hydrophobic chains: Hydrophobic functional groups are characterized by polar-free bonds and the inclusion of hydrophobic groups involves the attachment of alkaline chains (e.g. CH3, - C2H5) to the silicon surface.

- Effect on surface properties: Adding hydrophobic groups produces water-returning substances on the silicon surface and reduces its proximity to polar substances. This modification is useful in applications where it is necessary to return water or enhance



interaction with polar-free substances. For example, hydrophobic surface modification can be used for the efficient separation of polar-free analytics in chromatography.

In both cases, the choice of functional groups depends on specific application requirements. Superficial modifications are generally used in areas such as catalysis, sensors, chromatography and biomaterials, where controlling surface properties is crucial for performance optimization. The balance between hydrophilicity and hydrophobicity can be finely adjusted based on the intended function of the material.

Surface area and hole size:

-Controllable hydrophilic-hydrophobic silica particles can be controlled during the synthesis of specific surface area and hole size. These features are crucial for applications such as adsorption and catalysis.

Surface Area: The specific surface area of silica particles refers to the total area of all external surfaces corresponding to the unit mass or size of the material. This is an important parameter in various applications, specifically adsorption and catalysis. The higher specific surface area refers to more available areas for interaction with other substances, which is favorable for processes such as adsorption, where the material holds and holds molecules on its surface.

Controlling surface area during synthesis is essential for adapting the material to specific applications. The surface area of silica particles can be influenced by factors such as particle size, morphology and presence of mesoporas or microporas.

Teshik Hajmi:

The size of the holes is another important property that can be controlled in the synthesis of silica particles. The holes are divided into different categories, depending on the size:

1. Microporas: holes with a diameter of less than 2 nm.

2. Mesopora: holes from 2 nm to 50 nm in diameter.

3. Macropores: holes larger than 50 nm in diameter.

Controlling the size of the hole is very important because it affects the entry of molecules into the inner surface of the material. For example, in adsorption processes, the size of adsorbate molecules must match the size of the hole to ensure an efficient adoption.

Attachments:

- Adsorption: Controlled surface area and holesize materials are used in adsorbation processes such as water purification, degassing and removal of pollutants from air and water.
- Catalysis: Catalytic reactions often involve the interaction of reagents with the catalyst surface. A higher specific surface area and

corresponding hole size can increase catalytic activity by providing more active areas and facilitating the dispersion of the reactants.

- Drug Delivery: Controlled hydrophilichydrophobic silicon particles with a specific surface area and a hole size can be used as carrier for drug delivery systems, allowing for therapeutics to be released under control.
- Sensors: The adaptation of the surface area of silica particles and the size of the hole is crucial in the development of sensors, where the interaction between the targeted analytical and material surface is crucial for detection.

The synthesis of hydrophilic-hydrophobic silicon particles with controllable surface area and hole size is a manifold approach for certain applications, especially for adapting materials to adsorption and catalysis. These properties play a key role in determining the material's effectiveness in various processes.

Particle size and morphology:

- The size and shape of silica particles can be developed according to certain applications. Smaller sizes of particles can be favorable for certain applications, while in others, larger particles can be given the advantage.

Contact Angle:

- The contact angle of the water droplets on the silica surface is the main indicator of the material's hydrophobicity or hydrophilicity. The controllable hydrophilic hydrophobic silicon needs to show the intermediate contact angle due to dual modification.

Chemical Stability:

- The chemical stability of the modified silicon is important for its operation in different environments. Under the established use conditions, the chemical must be resistant to degradation.

Thermal Stability:

- Silica should maintain its properties at a number of temperatures, depending on the intended application. This is especially important during high-temperature processes.

Dispersion and agglomeration:

- The ability of modified silicon to propacate in agglomerated environments is essential for applications such as coatings, compositions and suspensions.

Adsorption and absorption characteristics:

- Controllable hydrophilic-hydrophobic silicon can be designed for the adsorption or ingestion of certain substances. This is especially relevant in applications like drug delivery, where it has to be released under control.

Compatibility with polymers or other materials:



- If the modified silicone is intended for use in composite materials or polymer matrixes, its compatibility with these materials is crucial to achieve optimal performance.

Rheological Features:

- The rheological movement of silica suspensions or dispersions is important for applications such as ink, paint and coating. Modified silica should contribute to the necessary rheological features of the system.

It is important to note that specific physics and chemical properties can vary depending on the nature of the synthesis methods and functional groups used for modification. The sewing of these properties allows to find applications in areas such as drug delivery, coatings, catalysis and environmental adjustment to controlled hydrophilic-hydrophobic silicon.

Variations observed in physical and chemical properties have a significant impact on different applications. Adaptation of the surface properties of the cream opens up new ways to design materials with improved performance in a specific environment. Discussion explores the potential application of hydrophilic hydrophobic silicone in areas such as drug delivery, oil – water separation and catalysis.

## **CONCLUSIONS:**

This study provides valuable insights into the controlled hydrophilic-hydrophobic modification of silicon and its physico-chemical properties. Silica surfaces' ability to fine tune moisture behavior offers a versatile platform for designing materials with customized functions.

While this work provides a comprehensive overview, there is still room for further intelligence. Future studies may focus on optimizing synthesis methods, exploring new methods of surface modification and expanding the application of hydrophilic-hydrophobic silicon in emerging fields.

In summary, the controlled modification of silica surfaces offers a promising path to the development of material science. By understanding and manipulating the physicocial properties of hydrophilic-hydrophobic silicon, researchers can contribute to the development of innovative solutions in various industries

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